

# Hercynite–quartz-bearing granulites from Brejões Dome area, Jequié Block, Bahia, Brazil: Influence of charnockite intrusion on granulite facies metamorphism

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Received 1 June 2005; accepted 30 March 2006

Available online 30 June 2006

## Abstract

In the present study, we describe and discuss the geology of aluminous–magnesian granulites and associated garnet-bearing charnockitic granulite from the Brejões Dome area, located in the Jequié Block, part of the São Francisco Craton in Bahia, Brazil. Investigation of metamorphic mineral assemblages allows the evaluation of  $P$ – $T$  conditions for the formation of these rocks, and therefore to obtain constraints for the better understanding of the geological evolution of the area. We conclude that the rocks from the Brejões Dome area were formed under granulite facies conditions of low to intermediate pressure (5–8 kbar). Temperatures determined in samples of aluminous–magnesian granulites collected away from the dome are in the order of 850 °C, similar to those determined elsewhere in the southern part of the Itabuna–Salvador–Curaçá Orogen. However, samples of the same rock type collected close to the Brejões Dome are hercynite+quartz-bearing and record higher temperatures of about 900–1000 °C. It is suggested that the intrusion of the Brejões charnockite diapir was responsible for a local increase in temperature above the peak temperature of regional granulite metamorphism.

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**Keywords:** Granulite; Jequié Block; Paleoproterozoic; São Francisco Craton; Spinel+quartz; Ultrahigh-temperature

## 1. Introduction

Aluminous–magnesian granulites contain a wide variety of mineral assemblages that are sensitive to  $P$ – $T$  conditions of metamorphism, and commonly preserve microstructures which are helpful to determine the sequence of mineral reactions (Hensen, 1987; Waters, 1991).

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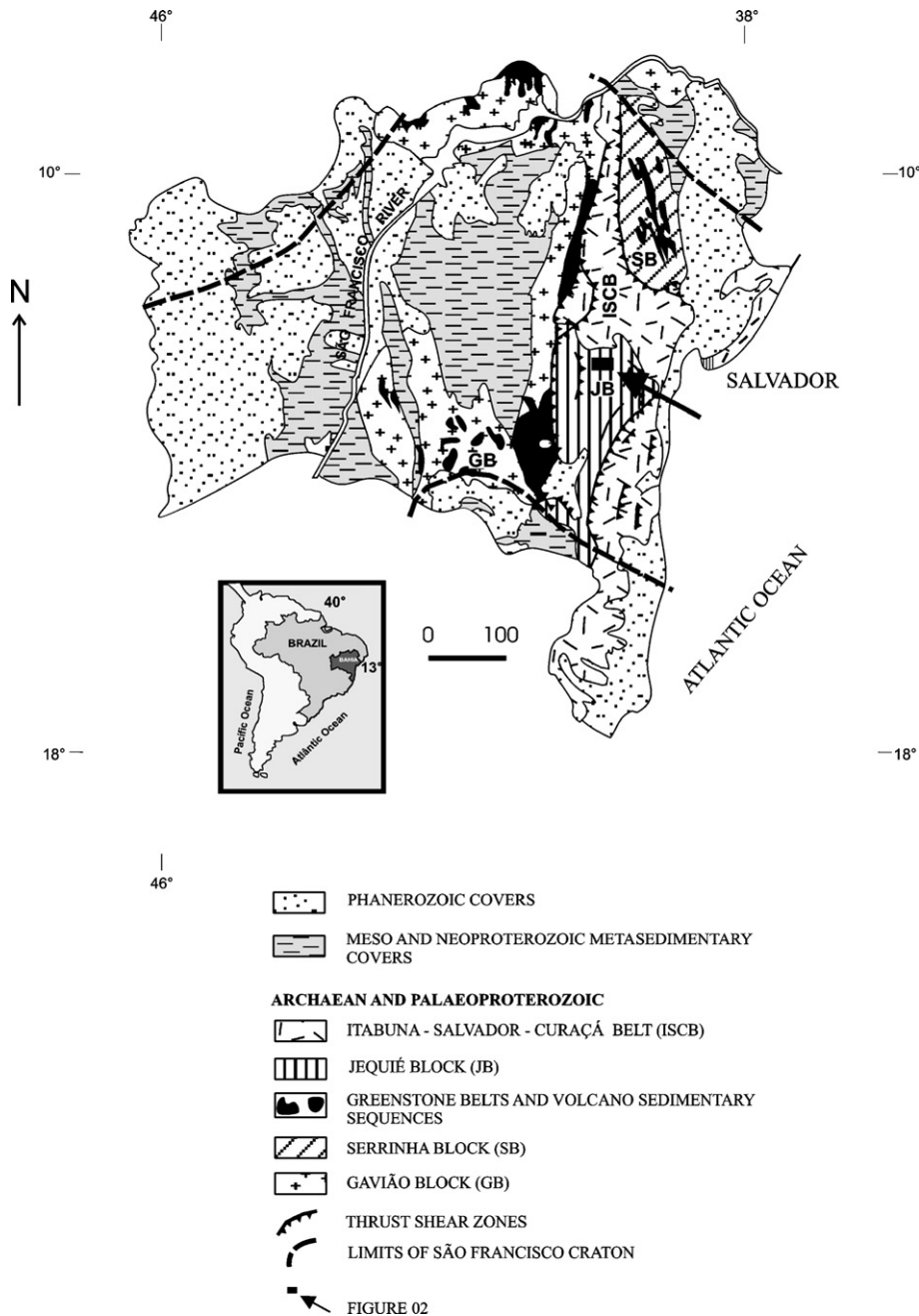


Fig. 1. Geotectonic domains in Bahia (simplified from Barbosa and Dominguez, 1996) with location of the study area.

Quartz–spinel and quartz–hercynite (Zn-poor spinel) assemblages are observed in aluminous–magnesian granulites from a number of areas on Earth. Well known examples are exposed in Antarctica (e.g. Ellis et al., 1980; Sandiford and Powell, 1986; Clarke et al., 1989), Norway (e.g. Tobi et al., 1985), India (e.g. Lal et al., 1987), Australia (e.g. Warren and Hensen, 1989), South Africa (e.g. Waters, 1989), Madagascar (e.g. Nicollet, 1990),

Algeria (e.g. Ouzegane and Boumaza, 1996), and Brazil (Moraes and Fuck, 2000; Moraes et al., 2002). Investigation of these areas and experimental data have shown that the hercynite+quartz association is formed at high temperature and low/intermediate pressure (Waters, 1991). Additionally, the Zn-poor hercynite+quartz paragenesis is of great interest since it reflects abnormal thermal conditions and an abnormal thermal gradient

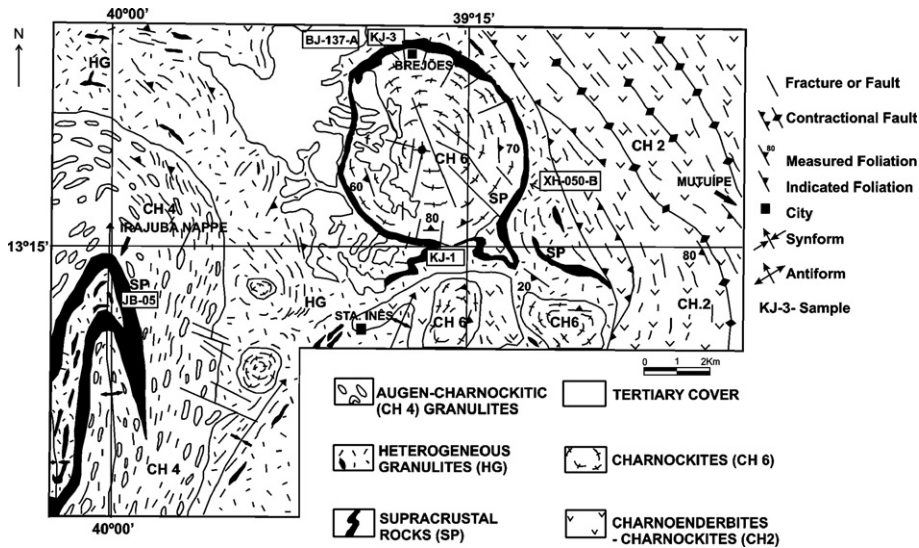


Fig. 2. Simplified geological map of the Brejões Dome area, Jequié Block, Bahia, Brazil (Barbosa et al., 2004).

within the continental crust. Therefore, the study of these granulites may lead to constrain the  $P$ – $T$  conditions and the geodynamic context in which they were formed.

In the present study, we describe and discuss the geology of hercynite–quartz-bearing aluminous–magnesian granulites and associated garnet–charnockitic granulites from the Brejões Dome area, in the Jequié Block, Bahia, eastern Brazil. Investigation of metamorphic mineral assemblages has allowed the evaluation of the  $P$ – $T$  conditions of formation of these rocks which enhances the understanding of the geologic evolution of this area.

## 2. Geological setting

The study area is located in the northern part of the Jequié Block, Bahia, eastern Brazil. Together with the archaean Gavião and Serrinha blocks, the Jequié Block is part of the basement framework of the São Francisco Craton (Almeida et al., 1976; Almeida, 1977) (Fig. 1). The archaean blocks were amalgamated at ca. 2.08 Ga during a paleoproterozoic collision (Barbosa and Sabaté, 2002, 2004), which gave birth to the high-grade Itabuna–Salvador–Curaçá belt.

The Jequié Block includes enderbitic–charnockitic granulites (CH2), some of which were dated at 2.7–2.8 Ga (Alibert and Barbosa, 1992). They are younger than a very heterogeneous association of orthoderived rocks including charnockitic–tonalitic granulites, mafic and felsic intercalations of varied composition which have  $T_{DM}$  Nd model ages between 2.90 and 3.25 Ga (Barbosa et al., 2004). This association is referred to as heterogeneous granulites (HG) in this paper. The HG are associated with

granulite facies supracrustal rocks (SP), comprising basaltic metavolcanic rocks, and several types of metasedimentary rocks, e.g. quartzite, iron formation, graphite-rich layers, and aluminous–magnesian granulites which locally contain the assemblage hercynite + quartz, object of this study (Fig. 2). All these rocks display partial melting features, such as migmatitic veins, and occasionally host garnet and cordierite-bearing anatectic granites.

The rocks of the Jequié Block were deformed by three continuous ductile deformation phases, contemporaneous with the paleoproterozoic high-grade metamorphism which affected the region at c.a. 2.1–2.0 Ga (Wilson, 1987; Ledru et al., 1994). The first phase (F1) is represented by large west-verging recumbent folds, with sub-horizontal axial planes and N–S fold axes, the most spectacular example of which is the Irajuba nappe (Fig. 2). The coaxial second phase (F2) comprises open folds in the Jequié Block and tight folds in the Itabuna–Salvador–Curaçá Belt, both with near vertical axial planes. The third phase (F3), well displayed within the Itabuna–Salvador–Curaçá Belt, is represented by vertical shear zones, which are sub-parallel to the upright axial planes of the tight F2 folds. The last phase gave origin to a vertical foliation with subhorizontal lineations. Kinematic indicators show that movement was top to northeast. Due to the presence of retrograde minerals such as hornblende and biotite formed after orthopyroxene in some transposition corridors, it is suggested that some parts of the high-grade region were reactivated during uplift of the granulite terrain (Barbosa, 1990).

Geothermobarometric data on the high-grade metamorphism in southern Bahia were obtained through

petrological investigation of rare garnet-bearing mafic granulites. Pressures of about 5–7 kbar and high temperatures between 850 and 870 °C were determined (Barbosa, 1986, 1990; Barbosa and Fonteilles, 1991).

In this work, previously obtained thermobarometric data will be compared with the results obtained for aluminous–magnesian granulite of the supracrustal rock layers (SP) and garnet-bearing charnockitic granulite from the HG which occur in the Brejões Dome area, in the northern part of the Jequié Block.

### 3. Geology of the Brejões Dome area

The HG (Fig. 2) are exposed in the central part of the study area, surrounding the Brejões, Santa Inês, and several other charnockite domes. This unit, with  $T_{DM}$  model ages of 2.9–3.2 Ga is essentially formed of an association of charnockitic–tonalitic granulites, with mafic and felsic intercalations of varied composition, equilibrated under granulite facies conditions (Barbosa et al., 2004). HG are usually greenish-gray, and are coarse- to medium-grained. They may be migmatitic, containing orthopyroxene-bearing leucosomes. Some of these rocks have garnet–orthopyroxene–plagioclase–quartz assemblages (Fig. 2). The garnet-bearing charnockitic granulite (sample XH-050-B), collected east of the Brejões Dome, is a dark-green banded rock, with centimetre-wide darker orthopyroxene-rich bands and lighter quartz–mesoperthite-dominated bands. Garnet crystals are dispersed within the rock, but tend to be more abundant in the dark bands.

The SP (Fig. 2) occur as enclaves or intercalations within HG. They constitute 20–50 cm to 1–5 m thick layers, concordant with the regional rock foliation (Barbosa, 1986, 1990). They sometimes form much larger layers, tens or hundreds of metres thick, as for example that which surrounds the Brejões Dome and in the Irajuba nappe (Fig. 2). The supracrustal rocks comprise mainly mafic granulites (formerly basalts and andesitic basalts), quartzite, iron formation, quartz–feldspar layers, aluminous–magnesian granulites that include hercynite–quartz-bearing granulite, and scattered graphite-rich layers. The aluminous–magnesian granulites are probably the result of metamorphism of pelitic rocks (Barbosa, 1990). In general, these rocks are medium to coarse-grained. The main outcrops are found within the valleys around the Brejões and Santa Inês domes (samples KJ-1, KJ-3, BJ-137-A) and in the Irajuba nappe (sample JB-05) in the western part of the study area (Fig. 2). The aluminous–magnesian granulites occur as metre to centimetre thick greenish-red layers, rich in pink garnet crystals, as well as cordierite+spinel clots, within a quartz–feldspar matrix. In some outcrops the aluminous–

magnesian granulites are intruded by undeformed garnet- and cordierite-bearing granites. Petrological and geochemical data indicate that these granites are S-type intrusions (Barbosa et al., 2004), and they are probably products of partial melting which occurred during peak-metamorphic conditions. They have been dated at around 2.0 Ga by the Pb–Pb evaporation method of single zircon crystal, and by the Th–U–Pb in-situ electron microprobe method on monazite (Barbosa et al., 2004).

The CH2 (charnoenderbitic–charnockitic granulites) (Fig. 2) exposed around Mutuipé, in the eastern part of the study area, tectonically overly HG and SP (Fig. 2). U–Pb zircon age determinations indicate that their protoliths are ca. 2.7–2.8 Ga old (Alibert and Barbosa, 1992). They are composed of mesoperthite (30–40%), quartz (30%), antiperthitic plagioclase (10–20%), hornblende (10–20%), orthopyroxene–clinopyroxene (2–5%); biotite, opaques, zircon and apatite are accessory minerals.

The CH4 (charnockitic augen-type granulites) (Fig. 2), also called the Irajuba type, occurs in the western part of the study area. In its central part, this coarse-grained intrusive body includes a large refolded belt of heterogeneous granulite surrounded by a thick layer of granulitic supracrustal (SP) rocks. The Irajuba charnockitic granulite is essentially composed of large, up to 4 cm mesoperthite porphyroclasts (40–50%), quartz (30–40%), antiperthitic plagioclase (5–10%); hornblende, orthopyroxene, and opaques are accessory minerals.

The CH6 (intrusive charnockite bodies) (Fig. 2) constitute the cores of the Brejões, Santa Inês and several other smaller, unnamed oval-shaped domes. They were dated at ca. 2.0 Ga (Barbosa et al., 2004). CH6 charnockite is formed of mesoperthite (30–40%), antiperthitic plagioclase (30%), quartz (20%), hornblende (10%), orthopyroxene (5%), and clinopyroxene (2–5%); opaque minerals (ilmenite, magnetite, pyrite), apatite, and zircon are accessory minerals.

Mineralogy of these rocks is fairly similar, however petrologic and geochemical studies led to identification of different types of parent magmas for these units. Granulite rocks CH2 and CH4 are mainly granodioritic to granitic, medium-K calc-alkaline suites. CH6 charnockites of Brejões and Santa Inês domes are more siliceous and are typically granite. Geochemical modelling shows that the syn-dome CH6 charnockites probably are products of melting of a CH2-type precursor (Barbosa et al., 2004).

### 4. Petrography

Samples analysed to constrain high-grade  $P$ – $T$  conditions recorded in the study area were collected from aluminous–magnesian granulites, exposed both away



Table 1  
Chemical data and mineral assemblages

Samples	Peak metamorphic phases											
	Grt1	Grt1 Al <sub>2</sub> O <sub>3</sub>	Opx1	Opx1Al <sub>2</sub> O <sub>3</sub>	Sill1	Crd1	Sp11	Bt1	Bt1Ti	Bt1F	Pl1	Mp1
KJ-3	0.37	21.19			X	0.80	0.34	0.80	0.28	0.00	X	
KJ-1	0.33	21.29			X	0.77	X	X			X	
BJ-137-A	0.38	21.67	0.64	6.04	X	0.82	X	0.74	0.34	0.02	X	X
JB-05	0.36	22.45	0.57	6.04	X	X		0.68	0.39	(na)	X	X
XH-050-B	0.16	20.89	0.38	2.55				0.63	0.28	0.00	X	

Samples	Retrograde phases									
	Opx2	Opx2Al <sub>2</sub> O <sub>3</sub>	Crd2	Sp12	Opx3	Opx3Al <sub>2</sub> O <sub>3</sub>	Crd3	Bt2	Ti	F
KJ-3			X					X		
KJ-1			0.75	0.21				X		
BJ-137-A	0.65	3.30	X		0.61	3.00	0.67	X		
JB-05			X					X	0.30	0.68
XH-050-B								0.60	0.27	0.00

All members refer to X<sub>mg</sub>, except Al<sub>2</sub>O<sub>3</sub> for orthopyroxene and garnet, Ti and F for biotite, and Na for plagioclase. Al<sub>2</sub>O<sub>3</sub> (wt.%), Ti, F and Na (a.p.f.u.) are average analyzed values.

from (sample JB-05) or close to (samples KJ-1, KJ-3, BJ-137-A) the contact with the Brejões Dome charnockite intrusion, and from garnet-bearing charnockitic granulite (sample XH-050-B) (Fig. 2).

#### 4.1. Aluminous–magnesian granulites

Samples KJ-1 and KJ-3 exhibit pervasive mylonitic foliation marked by quartz (Qtz, 35–40%) in ribbons (up to 12.0 mm) associated to elongate and almond-shaped grains of plagioclase (Pl1, 30–40%), garnet (Grt1, 8–10%), biotite (Bt1, 5%), sillimanite (Sill1, 3%), spinel (Sp11, 1–3%) and cordierite (Crd1, 10–15%) (Table 1). Ilmenite (1–2%), magnetite/pyrite (1–2%), rutile (1–2%), graphite (1%), monazite and zircon occur as inclusions and constitute accessory minerals. In samples BJ-137-A and JB-05, orthopyroxene (Opx1, 5%) and rare interstitial mesoperthite (Mp1) are observed in addition to the above-mentioned assemblage (Table 1).

Quartz (Qtz) occurs as oriented porphyroblasts (2.0–4.3 mm) and xenomorphic (0.3–1.2 mm) crystals that display undulating extinction, deformation bands and development of sub-grains with serrated to lobate boundaries with other quartz grains. Quartz also forms microstructures which bulge into neighbouring deformed garnet porphyroblasts (Fig. 3a, b). These mylonitic microstructures attest to deformation under granulite-facies metamorphic conditions. Spinel is present as lamellae and globules (Sp11) inside cordierite, sometimes forming symplectites (sample KJ-3, Fig. 3a, b), or as idiomorphic to hypidiomorphic grains (Sp12) between 0.3 and 0.6 mm long in-

cluded in quartz. In the latter case it may have well-developed crystallographic faces against quartz, as in samples KJ-1 and KJ-3 collected at the contact with CH6 charnockites of the Brejões and Santa Inês domes (Fig. 3c), which suggests equilibrium conditions, and permits to constrain minimum *P–T* conditions for the rocks. Plagioclase (Pl1) forms hypidiomorphic porphyroblasts up to 3 mm long with quartz and biotite inclusions and polygonal aggregates of idiomorphic crystals up to 0.6 mm long. Most of the porphyroblasts have K–feldspar lamellae, and are therefore antiperthitic plagioclase (Antp) with albite twinning which is sometimes bent. Garnet (Grt1) occurs as xenomorphic and hypidiomorphic crystals (0.3–1.0 mm), and porphyroblasts (1.0–3.0 mm) with straight or lobate contacts with cordierite and quartz (Fig. 3a, d). The porphyroblasts have small inclusions of quartz, plagioclase, ilmenite and biotite (Fig. 3a, b, d). Biotite (Bt1) has reddish-brown pleochroism, and occurs as very small (0.1 mm) inclusions within quartz, garnet and plagioclase (Fig. 3d, e), as well as oriented inclusions up to 1.0 mm long in orthopyroxene (Fig. 3f). Sillimanite (Sill1) occurs as small inclusions (<0.1 mm) in cordierite (Fig. 3a, b), or as oriented prisms in contact with quartz or biotite. Cordierite (Crd1) forms porphyroblasts and moats around garnet (Fig. 3a, b), in grains which reach 3.0 mm, have polygonal and lobate contacts, and show polysynthetic twinning. Another generation of cordierite (Crd2) forms coronae around spinel in contact with quartz (Fig. 3c, g). In the first type, pinitization is absent or limited to micro-channels which cut the grains, while in the second type it is more widespread. The third type of cordierite (Crd3) occurs in

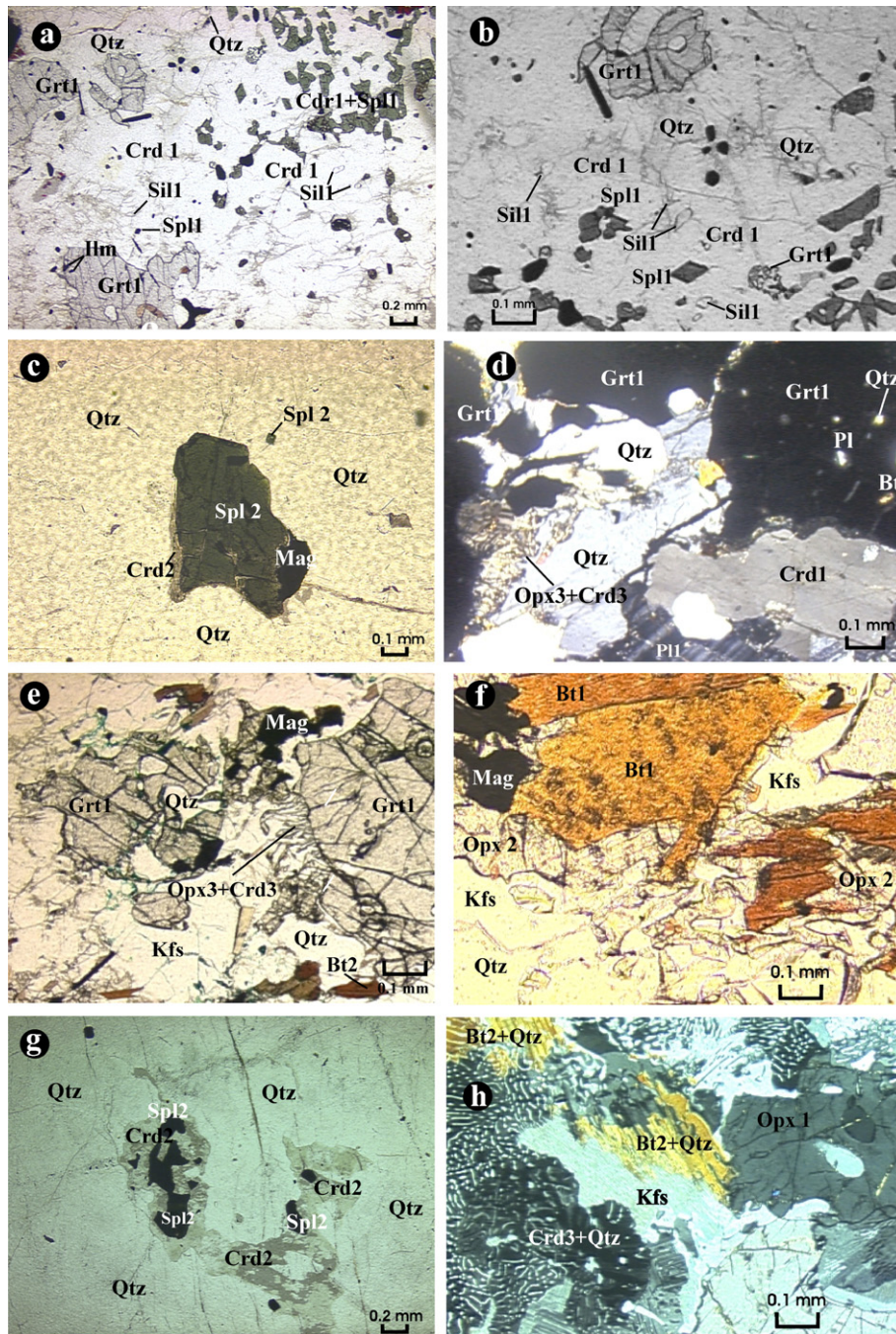


Fig. 3. (a) Crd1 + Spl1 symplectite as the product of reaction:  $\text{Grt1} + \text{Sil1} (\pm \text{Qtz}) = \text{Crd1} + \text{Spl1}$ . Hercynite–quartz-bearing granulite, sample KJ-3. (b) Detail of photo 3a. Please note that sillimanite and quartz were virtually consumed in the reaction  $\text{Grt1} + \text{Sil1} (\pm \text{Qtz}) = \text{Crd1} + \text{Spl1}$ . Hercynite–quartz-bearing granulite, sample KJ-3. (c) Hercynite crystal (Spl2) with straight borders within quartz. Hercynite–quartz-bearing granulite, sample KJ-1. (d) Biotite, quartz and plagioclase within garnet (Grt1), and reaction  $\text{Grt1} + \text{Qtz} = \text{Opx3} + \text{Crd3}$ . Hercynite–quartz-bearing granulite, sample KJ-1. (e) Orthopyroxene (Opx3) + cordierite (Crd3) symplectite formed around garnet, according to decompression reaction  $\text{Grt1} + \text{Qtz} = \text{Opx3} + \text{Crd3}$ . Hercynite–quartz-bearing granulite, sample BJ-137-A. (f) Orthopyroxene (Opx2) + mesoperthite (Kfs) formed according to dehydration reaction  $\text{Bt1} + \text{Qtz} = \text{Opx2} + \text{Kfs} + \text{L}$ . Hercynite–quartz-bearing granulite, sample BJ-137-A. (g) Hercynite crystals (Spl2) within quartz, surrounded by cordierite (Crd2) reaction coronae, according to retrograde reaction  $\text{Spl2} + \text{Qtz} = \text{Crd2}$ . Hercynite–quartz-bearing granulite, sample KJ-1. (h) Symplectite of Bt2 + Qtz resulting from retrograde reaction  $\text{Opx1} + \text{Kfs} + \text{L} = \text{Bt2} + \text{Qtz}$ . Hercynite–quartz-bearing granulite, sample KJ-3.

Table 2  
Representative chemical composition of garnet

Samples	KJ-3	KJ-3	KJ-3	KJ-1	KJ-1	KJ-1	BJ-137-A	BJ-137-A	JB-05	JB-05	XH-050-B	XH-050-B
Analysis no.	17	24	25	6	12	20	4	5	170	171	9	19
Remarks	Rim	Core	Rim	Rim	Core	Rim	Core	Core	Core	Core	Core	Rim
SiO <sub>2</sub>	38.98	38.89	38.86	38.25	38.44	38.59	38.92	39.09	38.58	38.35	37.48	37.97
TiO <sub>2</sub>	0.07	0.14	0.07	0.07	0.04	0.02	0.08	0.06	0.06	0.07	0.05	0.02
Al <sub>2</sub> O <sub>3</sub>	20.94	21.31	21.38	21.64	21.12	21.12	21.9	21.44	22.64	22.26	20.87	20.92
Cr <sub>2</sub> O <sub>3</sub>	0.15	0.08	0.16	0.09	0.19	0.13	0.07	0.09	0.17	0.23	0.05	0.00
FeO	27.45	27.44	28.34	29.00	29.51	30.38	26.26	25.97	27.57	27.88	32.37	33.76
MnO	0.47	0.42	0.52	1.58	1.88	1.71	0.30	0.40	1.32	1.40	1.13	1.30
MgO	9.33	9.49	9.45	8.12	7.89	7.37	9.96	9.87	8.84	9.04	4.12	3.36
CaO	1.23	1.6	1.07	1.23	1.12	1.24	1.25	1.25	1.67	1.23	2.16	1.97
Na <sub>2</sub> O	0.05	0.04	0.00	0.02	0.00	0.01	0.01	0.01	0.01	0.01	0.02	0.02
K <sub>2</sub> O	0.04	0.03	0.00	0.03	0.03	0.01	0.03	0.00	0.00	0.00	0.00	0.00
F	0.36	0.00	0.00	0.09	0.00	0.00	0.31	0.36	0.00	0.00	0.00	0.00
Sum	99.05	99.44	99.85	100.12	100.22	100.58	99.09	98.24	100.86	100.47	98.25	99.32

Structural formulae based on 12 oxygens

Si	2.99	3.01	3.00	2.97	2.99	3.00	3.06	3.05	2.99	2.99	3.04	3.02
Al <sup>IV</sup>	0.01	0.00	0.00	0.03	0.01	0.00	0.00	0.00	0.01	0.01	0.00	0.00
Al <sup>VI</sup>	1.94	1.94	1.95	1.95	1.93	1.94	1.98	1.97	2.02	1.99	1.99	2.01
Ti	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.00	0.00	0.01	0.00
Cr	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.00
Fe <sup>3+</sup>	0.06	0.03	0.03	0.06	0.06	0.05	0.00	0.00	0.00	0.00	0.00	0.00
Fe <sup>2+</sup>	1.76	1.74	1.79	1.82	1.85	1.93	1.68	1.68	1.74	1.77	2.19	2.30
Mn	0.04	0.03	0.03	0.10	0.12	0.11	0.02	0.02	0.08	0.09	0.07	0.06
Mg	1.09	1.09	1.09	0.94	0.91	0.85	1.14	1.15	0.99	1.02	0.49	0.41
Ca	0.10	0.13	0.09	0.10	0.09	0.10	0.10	0.10	0.13	0.10	0.18	0.17
Na	0.01	0.01	0.00	0.01	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.01
K	0.01	0.01	0.00	0.01	0.01	0.01	0.00	0.00	0.00	0.00	0.00	0.00
X <sub>Mg</sub>	0.38	0.38	0.37	0.34	0.33	0.31	0.38	0.38	0.38	0.34	0.18	0.15
Al	74.00	68.70	68.00	61.30	62.10	64.30	57.10	56.70	58.90	59.30	74.10	78.00
Py	16.70	23.60	23.80	31.70	30.60	28.50	38.60	38.80	33.60	34.30	16.80	13.80
Sp	2.80	4.40	4.60	3.50	4.10	3.80	0.70	0.90	2.90	3.00	2.60	2.30
Gr	6.30	1.30	0.40	0.00	0.00	0.50	3.00	3.00	3.90	2.40	6.00	5.70
Ad	0.00	1.00	2.80	3.40	3.10	2.40	0.00	0.00	0.00	0.00	0.00	0.00
UV	0.20	0.70	0.40	0.10	0.00	0.40	0.20	0.30	0.50	0.70	0.02	0.00

XH-050-B is garnet-bearing charnockitic granulite sample and the others are aluminous–magnesian granulites.

symplectites with orthopyroxene (Opx3) around garnet (Grt1) (Fig. 3d, e) or in symplectites with quartz close to orthopyroxene porphyroblasts (Opx1) (Fig. 3h). Orthopyroxene (Opx1) occurs as oriented xenomorphic to hypidiomorphic crystals with dimensions of 0.3–1.0 mm (Fig. 3h), some of them with biotite inclusions. Orthopyroxene also forms symplectitic coronae (Opx2) with K–feldspar (Kfs) surrounding biotite (Bt1) (Fig. 3f) in sample BJ-137-A at the contact with the charnockite domes. A third generation of orthopyroxene occurs as lamellae and globules (Opx3) in cordierite-host symplectites (Fig. 3d, e).

In relation to the accessory phases only magnetite appears in modal proportion greater than 1%. It is in contact with spinel, garnet and biotite (Fig. 3c, e, f). Additional opaque minerals (ilmenite, graphite) are scattered throughout the rock or sometimes included in garnet.

Monazite and zircon are readily recognized as inclusions in cordierite and biotite due to the yellowish pleochroic haloes produced in their host-grains.

Microstructural relationships indicate that the peak metamorphic mineral paragenesis of the aluminous–magnesian granulites is composed of Qtz+Sp1+P11+Grt1+Sil1+Bt1. Retrograde granulite-metamorphic paragenesis are seen in interstitial positions between peak metamorphic minerals and they are formed of: (i) a second generation of cordierite (Crd2) at the contact between hercynite and quartz (Fig. 3c); (ii) symplectitic coronae of cordierite (Crd3) with a third generation of orthopyroxene (Opx3) at the contact between garnet and quartz (Fig. 3d, e); (iii) a second type of biotite (Bt2) constitutes intergrowths with quartz in coronae around orthopyroxene (Fig. 3h) and opaque minerals.



Table 3  
Representative chemical composition of orthopyroxene

Samples	BJ-137 A	BJ-137 A	BJ-137 A	BJ-137-A	BJ-137-A	JB-05	XH-050-B	XH-050-B	XH-050-B
Analysis no.	2	3	11	14	17	1	1	2	4
Remarks	Opx1(Core)	Opx1(Core)	Opx2(Core)	Opx3(Core)	Opx3(Rim)	Opx1(Core)	Opx1(Core)	Opx1(Core)	Opx1(Rim)
SiO <sub>2</sub>	48.96	49.4	50.91	50.92	50.93	48.77	47.84	48.17	48.34
TiO <sub>2</sub>	0.13	0.18	0.04	0.14	0.15	0.15	0.15	0.13	0.16
Al <sub>2</sub> O <sub>3</sub>	5.67	6.42	3.30	3.00	3.00	6.04	2.60	2.63	2.44
Cr <sub>2</sub> O <sub>3</sub>	0.19	0.04	0.00	0.39	0.40	0.26	0.00	0.01	0.00
FeO	21.33	20.90	22.03	22.82	22.99	26.17	35.02	34.65	34.4
MnO	0.22	0.06	0.30	0.39	0.40	0.43	0.3	0.35	0.35
MgO	20.86	21.48	22.10	20.05	19.93	18.52	12.23	12.44	13.11
CaO	0.19	0.11	0.09	0.10	0.09	0.10	0.32	0.30	0.26
Na <sub>2</sub> O	0.02	0.03	0.04	0.06	0.07	0.01	0.02	0.03	0.00
K <sub>2</sub> O	0.00	0.00	0.01	0.04	0.04	0.00	0.00	0.03	0.01
F	0.04	0.16	0.00	0.27	0.20				
Sum	97.61	98.78	98.82	98.18	98.20	100.45	98.48	98.74	99.07

*Structural formulae based on 12 oxygens*

Si	1.86	1.84	1.91	1.94	1.91	0.16	1.93	1.93	1.95
Al <sup>IV</sup>	0.13	0.15	0.09	0.05	0.09	0.10	0.07	0.06	0.05
Al <sup>VI</sup>	0.12	0.13	0.05	0.10	0.07	0.01	0.05	0.06	0.02
Ti	0.00	0.00	0.00	0.00	0.01	0.01	0.00	0.00	0.01
Cr	0.00	0.00	0.00	0.00	0.01	0.04	0.00	0.00	0.00
Fe <sup>3+</sup>	0.00	0.01	0.03	0.00	0.01	0.01	0.01	0.00	0.02
Fe <sup>2+</sup>	0.67	0.64	0.65	0.72	0.72	0.78	1.17	1.16	1.14
Mn	0.00	0.00	0.01	0.00	0.01	0.01	0.01	0.01	0.01
Mg	1.18	1.19	1.23	1.15	0.16	1.04	0.73	0.74	0.79
Ca	0.00	0.00	0.00	0.00	0.00	0.00	0.01	0.01	0.01
Na	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.00
K	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
F	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
X <sub>Mg</sub>	0.64	0.65	0.65	0.61	0.61	0.57	0.38	0.39	0.52

XH-050-B is garnet-bearing charnockitic granulite sample and the others are aluminous–magnesian granulites.

#### 4.2. Garnet-bearing charnockitic granulite

Sample XH-050-B is constituted of mesoperthite (Mp1, 40–50%), quartz (Qtz, 30%), antiperthitic plagioclase (Pl1, 10%), orthopyroxene (Opx1, 10%), sometimes transformed into bastite, garnet (Grt1, 5–10%) and biotite (Bt1, 2–5%). Mesoperthite (Mp1) occurs as porphyroblasts (0.8–3.0 mm) with lobate to straight contacts with garnet and plagioclase. Myrmekite is commonly developed at its contact with plagioclase. Quartz (Qtz1) occurs as xenomorphic porphyroblasts (1.0–2.5 mm) that are fractured and exhibit strong undulating extinction and deformation bands. The porphyroblasts show lobate boundaries against the other grains. Antiperthitic plagioclase (Pl1) forms prismatic crystals (0.6–3.0 mm) with curved albite twins. The larger crystals contain inclusions of garnet and biotite and show K–feldspar lamellae. Orthopyroxene (Opx1) constitutes xenomorphic crystals (<0.5 mm), most of them altered to bastite. Orthopyroxene (Opx1) also forms coronae together with plagioclase (Pl1) around garnet megacrystals. Garnet (Grt1) forms idio-

morphic to hypidiomorphic porphyroblasts, 0.7 to 2.5 mm long, with borders resorbed by mesoperthite, plagioclase and quartz. The porphyroblasts contain inclusions of plagioclase, biotite and quartz. Biotite (Bt1) occurs as dark-red porphyroblasts that are dispersed among the other main mineral phases. Its cleavage is always oriented parallel to the rock foliation. A second generation of biotite (Bt2) is formed from orthopyroxene and garnet, and occurs in biotite–quartz symplectites, as well as flakes also oriented parallel to the rock foliation. Ilmenite, pyrite, apatite, zircon, and monazite are accessory phases. Myrmekite is also found along contacts between plagioclase, mesoperthite and quartz in these metaplutonic rocks re-equilibrated in granulite facies.

## 5. Mineral chemistry

### 5.1. Analytical methods

Mineral compositions were obtained at Paris VI University, France using the Camebax SX 100 electron



microprobe. Operating conditions were 15 kV accelerating voltage and 10 nA specimen current and 1–2 μm beam diameter. Natural silicates and synthetic oxides were used as standards for all elements. Representative analytical results of the main mineral phases of garnet-bearing charnockitic granulite (HG) and aluminous–magnesian granulite samples (SP) are shown in Tables 1–7. Structural formulae were calculated with the Norm Version 4.0 Program of UIm (1993). Abbreviations follow Kretz (1983), plus Mp for mesoperthite.

5.2. Aluminous–magnesian granulites

Garnet from aluminous–magnesian granulites displays small chemical variations among grains. It is relatively rich in the almandine component (about 60%), but it is richer in pyrope component (around 30–35%) than garnet (Grt1) crystals from the garnet-bearing charnockitic granulite. The grossular and spessartite end-

members are present in low concentrations (<5%) (Fig. 4, Table 2). Microprobe profiles across two garnet porphyroblasts from samples KJ-1 and KJ-3 reveal slight zoning of iron and magnesium contents. In the garnet porphyroblast from sample KJ-1 there is iron enrichment and magnesium depletion from core to rim, due to the take-up of Mg in the formation of the cordierite envelope. Mg and Fe zoning in the garnet porphyroblast from sample KJ-3 repeats this pattern. Ca is richer in the core than at the rim. This feature can be attributed to the formation of plagioclase around garnet, in which plagioclase took up Ca from the garnet rims.

Orthopyroxene in sample BJ-137-A occurs in three generations that are different in chemistry: Opx1 is richer in aluminium (≈ 6.0 wt.% Al<sub>2</sub>O<sub>3</sub>) than Opx2 (≈ 3.5 wt.% Al<sub>2</sub>O<sub>3</sub>), while Opx3 with ≈ 3.0 wt.% Al<sub>2</sub>O<sub>3</sub> has higher contents of Fe than Opx1 and Opx2 (Fig. 5, Table 3). The low Al<sub>2</sub>O<sub>3</sub> concentrations of around 6.0 wt.% in Opx1 stand out, being lower than those found in other

Table 4  
Representative chemical composition of biotite

Samples	KJ-3	KJ-3	BJ-137-A	BJ-137-A	JB-05	JB-05	XH-050-B	XH-050-B
Analysis no.	29	30	6	16	173	172	18	20
Remarks	Bt1(Core)	Bt1(Core)	Bt1(Core)	Bt1(Core)	Bt1(Core)	Bt2(Core)	Bt1(Core)	Bt2(Core)
SiO <sub>2</sub>	37.59	37.27	37.98	37.88	37.24	39.15	37.16	38.47
TiO <sub>2</sub>	5.12	5.73	5.13	6.61	4.50	3.94	4.84	4.58
Al <sub>2</sub> O <sub>3</sub>	15.9	15.55	14.98	14.23	15.65	16.63	13.36	12.38
Cr <sub>2</sub> O <sub>3</sub>	0.30	0.16	0.28	0.35	0.51	0.41	0.01	0.08
FeO	7.53	7.97	7.04	11.37	15.32	9.15	14.91	16.56
MnO	0.00	0.05	0.00	0.27	0.02	0.03	0.14	0.00
MgO	17.86	17.62	17.84	13.72	14.98	18.31	14.11	13.53
CaO	0.03	0.00	0.03	0.00	0.04	0.02	0.00	0.00
Na <sub>2</sub> O	0.42	0.37	0.42	0.05	0.12	0.08	0.12	0.09
K <sub>2</sub> O	9.31	9.52	8.93	9.51	9.35	9.27	9.77	9.80
H <sub>2</sub> O	2.98	3.26	3.29	3.63	2.04	2.11	3.99	3.56
F	2.39	1.76	2.44	2.63			0.00	0.00
Sum	99.56	99.38	98.36	100.25	99.8	99.1	98.41	99.05

Structural formulae based on 11 oxygens

Si	2.65	2.63	2.80	2.86	2.73	2.82	2.88	2.92
Al <sup>IV</sup>	1.22	1.19	1.19	1.14	1.26	1.17	1.11	1.07
Al <sup>VI</sup>	0.05	0.10	0.17	0.19	0.09	0.24	0.11	0.16
Ti	0.27	0.30	0.30	0.39	0.39	0.30	0.28	0.27
Cr	0.01	0.01	0.02	0.02	0.02	0.01	0.00	0.01
Fe <sup>3+</sup>	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Fe <sup>2+</sup>	0.44	0.47	0.45	0.75	1.06	0.55	0.97	1.01
Mn	0.00	0.01	0.00	0.02	0.00	0.00	0.01	0.00
Mg	1.88	1.85	2.06	1.62	1.48	1.97	1.63	1.55
Ca	0.01	0.00	0.00	0.00	0.05	0.00	0.00	0.00
Na	0.06	0.05	0.06	0.01	0.01	0.06	0.02	0.01
K	0.84	0.85	0.88	0.96	0.96	0.88	0.97	0.97
F	0.00	0.00	0.02	0.02	0.30	0.68	0.00	0.00
OH	0.74	0.81	2.00	2.00	2.00	2.00	2.00	2.00
X <sub>Mg</sub>	0.81	0.80	0.81	0.68	0.58	0.78	0.63	0.60

XH-050-B is garnet-bearing charnockitic granulite sample and the others are aluminous–magnesian granulites.

Table 5  
Representative chemical composition of cordierite and sillimanite

Samples	KJ-3	KJ-3	KJ-1	KJ-1	KJ-1	BJ-137-A	BJ-137-A	BJ-137-A	KJ-3
Analysis no.	13	14	31	32	36	19	9	13	1
Remarks	Crd1(Core)	Crd1(Core)	Crd1(Rim)	Crd1(Rim)	Crd2(around Hc)	Crd1(Core)	Crd3/Opx3	Crd3/Opx3	Sill
SiO <sub>2</sub>	48.97	49.34	49.37	49.42	48.79	48.81	49.83	49.83	36.65
TiO <sub>2</sub>	0.00	0.01	0.02	0.01	0.00	0.00	0.00	0.00	0.00
Al <sub>2</sub> O <sub>3</sub>	33.15	32.75	33.36	33.56	33.53	32.42	33.30	33.03	61.12
Cr <sub>2</sub> O <sub>3</sub>	0.05	0.07	0.04	0.15	0.00	0.12	0.00	0.00	0.08
FeO	4.91	4.99	5.32	5.31	6.07	4.02	4.70	4.91	0.21
MnO	0.10	0.00	0.00	0.15	0.10	0.01	0.00	0.00	0.00
MgO	10.90	10.71	10.25	9.98	9.86	10.76	9.80	10.15	0.04
CaO	0.00	0.00	0.00	0.00	0.03	0.00	0.01	0.01	0.02
Na <sub>2</sub> O	0.03	0.03	0.13	0.11	0.09	0.06	0.06	0.05	0.03
K <sub>2</sub> O	0.00	0.01	0.01	0.03	0.02	0.03	0.00	0.00	0.01
F	0.00	0.29	0.07	0.00	0.14	0.14	0.09	0.09	0.00
Sum	98.11	98.2	98.57	98.72	98.63	96.37	97.79	98.07	98.16
<i>Structural formulae based on 18 oxygens (Crd); 20 oxygens (Sil)</i>									
Si	4.97	5.02	5.00	4.99	4.96	5.03	5.04	4.99	4.03
Al	3.97	3.93	3.98	4.00	4.02	3.94	3.97	3.98	7.93
Ti	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Cr	0.01	0.01	0.00	0.01	0.00	0.01	0.00	0.00	0.01
Fe <sup>2+</sup>	0.42	0.42	0.45	0.45	0.51	0.35	0.62	0.52	0.02
Mn	0.01	0.00	0.00	0.01	0.01	0.00	0.00	0.00	0.00
Mg	1.65	1.63	1.55	1.50	1.49	1.65	1.20	1.21	0.01
Ca	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.00	0.00
Na	0.01	0.01	0.02	0.02	0.02	0.01	0.01	0.01	0.01
K	0.00	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.00
F	0.00	0.09	0.02	0.00	0.04	0.00	0.00	0.00	0.00
X <sub>Mg</sub>	0.80	0.80	0.77	0.77	0.75	0.82	0.66	0.67	0.33

All samples are aluminous–magnesian granulites.

aluminous–magnesian granulites in the rest of the world, such as those of the Central Limpopo Belt, which have Al<sub>2</sub>O<sub>3</sub> contents around 8.0 wt.% (Harris and Holland, 1984a,b), those of North-Central Madagascar with up to 9.0 wt.% of Al<sub>2</sub>O<sub>3</sub> (Gonçalves et al., 2004), and those of the Central Highland Complex whose orthopyroxene contains between 8.0 and 12.5 wt.% Al<sub>2</sub>O<sub>3</sub> (Sajeev and Osanai, 2004). The Al<sub>2</sub>O<sub>3</sub> contents in orthopyroxene seem to reflect those of the metapelite protholiths which are poor in alumina in the studied region (Barbosa, 1986). The molar concentrations of magnesium in the studied orthopyroxene (X<sub>Mg</sub>=0.60) are also lower than those in the three cited areas, which lie between 0.70 and 0.80.

Biotite occurs in two distinct generations (Bt1, Bt2) in samples KJ-3, BJ-137-A and JB-5. Peak/close-to-peak biotite (Bt1) is richer in magnesium (X<sub>Mg</sub> ≈ 0.81) than retrograde biotite (Bt2 X<sub>Mg</sub> ≈ 0.78) in symplectites with quartz (Tables 1, 4). Using the classification of Hormann et al. (1980), Bt1 is phlogopite–meroxene, whereas Bt2 is meroxene. Titanium contents of Bt1 vary between 5% and 6% in samples KJ-3, BJ-137-A from close to the Brejões Dome to around 4% in sample JB-5, away from

the dome. Fluorine contents are also higher (1.76–2.63 wt.%) in Bt1 than in Bt2, which has insignificant or zero concentrations.

Cordierite from different samples (Tables 1, 5) is present in three generations, Crd1, Crd2, and Crd3, which have decreasing X<sub>Mg</sub> ratios, in Crd1 between 0.77 and 0.82, in Crd2 around 0.75, and in Crd3, 0.66–0.67. The high wt.% totals in the range 96.37–98.7 indicate that only small quantities of fluid species such as H<sub>2</sub>O and CO<sub>2</sub> are present in the structural channels.

Spinel occurs in symplectites with cordierite (Spl1), or in direct contact with quartz (Spl2). The two types show important differences in composition (Fig. 6, Tables 1, 6). Both spinel generations are hercynite-rich solid solutions, with X<sub>Mg</sub> ratios varying from 0.33 to 0.34 for Spl1, and from 0.19 to 0.23 for Spl2. Cr<sub>2</sub>O<sub>3</sub> contents vary from 0.14 to 0.15 wt.% for Spl1, and from 2.32 to 4.73 wt.% for Spl2. ZnO contents vary from 0.76 to 0.82 wt.% for Spl1, and from 1.96 to 2.75 wt.% for Spl2.

Plagioclase compositions are variable even in the aluminous–magnesian granulites. In KJ-3 the Ab content lies between 61 and 66 wt.%, while in JB-5 it is around

50 wt.%. The potash feldspar is essentially orthoclase (Table 7). Ilmenite has TiO<sub>2</sub> contents of ca. 54% and FeO contents of ca. 46%. Magnetite, often associated with spinel, has an X<sub>Fe</sub> ratio (Fe<sup>+3</sup>/Fe<sup>+3</sup>+Fe<sup>+2</sup>) of 0.66.

### 5.3. Garnet-bearing charnockitic granulite

Garnet (Grt1) grains are unzoned and rich in almandine (74–75%) and pyrope (12–16%), and poor in spessartine (2.5–3%) (Fig. 4, Table 2). The almandine component is more abundant than in garnet from the aluminous–magnesian granulites, in which pyrope and spessartine are more abundant.

Orthopyroxene (Opx1) is ferrosilite-rich (X<sub>Mg</sub>=0.38, Fig. 5, Table 3) and it appears as 2–5 mm porphyroblastic crystals with biotite and opaque mineral inclusions. The aluminium contents (≈2.6 wt.%) are significantly lower than those found in the aluminous–magnesian granulites (≈6.0 wt.%).

Biotite (Bt1) is richer in Fe (15–16 wt.%) than biotite in aluminous–magnesian granulites (≈7–11 wt.%), while Al<sub>2</sub>O<sub>3</sub> (≈13 wt.%) and TiO<sub>2</sub> (4.6–4.8 wt.%) contents are lower (Table 4). Antiperthitic plagioclase (Pl1) and mesoperthite (Mp1) are andesine and orthoclase, respectively (Table 7).

## 6. Evolution of mineral assemblages

### 6.1. Aluminous–magnesian granulites

In the aluminous–magnesian granulites most of the peak metamorphic phases (Opx1+Bt1+Grt1+Sill+Qtz) are partially replaced. Greenish spots, oriented parallel to banding or foliation, are found in several outcrops. They are composed of cordierite (Crd1) plus spinel (Spl1) symplectites which surround corroded garnet grains. These microstructures are similar to those described in rocks from Forefinger Point, Napier Complex in Antarctica (Harley

Table 6  
Representative chemical composition of spinel

Samples	KJ-3	KJ-3	KJ-3	KJ-1	KJ-1	KJ-1	KJ-1	KJ-1
Analysis no.	3	5	7	1	2	34	43	45
Remarks	Spl1/Crd1 (Core)	Spl1/Crd1 (Core)	Spl1/Crd1 (Core)	Spl2/Qz (Core)	Spl2/Qz (Core)	Spl2/Qz (Core)	Spl2/Qz (Core)	Spl2/Qz (Rim)
SiO <sub>2</sub>	0.00	0.03	0.00	0.00	0.03	0.00	0.02	0.03
TiO <sub>2</sub>	0.02	0.05	0.09	0.01	0.05	0.06	0.03	0.00
Al <sub>2</sub> O <sub>3</sub>	59.16	59.18	59.42	54.75	55.48	51.27	55.36	54.81
Cr <sub>2</sub> O <sub>3</sub>	0.15	0.15	0.14	2.68	2.45	4.73	2.32	2.97
FeO	30.58	30.7	29.46	34.03	34.03	35.45	34.00	34.13
MnO	0.10	0.04	0.06	0.23	0.16	0.23	0.30	0.23
MgO	8.08	8.21	8.25	5.33	5.20	4.13	4.94	4.75
CaO	0.00	0.00	0.00	0.00	0.00	0.01	0.06	0.00
Na <sub>2</sub> O	0.03	0.03	0.05	0.06	0.03	0.11	0.07	0.05
K <sub>2</sub> O	0.00	0.00	0.01	0.01	0.00	0.00	0.02	0.00
ZnO	0.80	0.82	0.76	2.07	1.96	2.75	2.15	2.09
F	0.33	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Sum	99.25	99.21	98.24	99.17	99.39	98.74	99.27	99.06

### Structural formulae based and 4 oxygens

Si	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al	1.94	1.94	1.96	1.87	1.89	1.87	1.89	1.88
Ti	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.00
Cr	0.01	0.01	0.01	0.06	0.05	0.09	0.05	0.07
Fe <sup>3+</sup>	0.05	0.05	0.04	0.07	0.05	0.13	0.05	0.05
Fe <sup>2+</sup>	0.66	0.65	0.65	0.76	0.77	0.79	0.77	0.78
Mn	0.01	0.00	0.01	0.01	0.01	0.01	0.01	0.01
Mg	0.33	0.34	0.34	0.23	0.22	0.19	0.21	0.21
Ca	0.00	0.00	0.00	0.00	0.00	0.01	0.01	0.00
Na	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
K	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.00
F	0.23	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Zn	0.13	0.13	0.12	0.35	0.33	0.48	0.36	0.35
X <sub>Mg</sub>	0.33	0.34	0.34	0.23	0.22	0.19	0.21	0.21

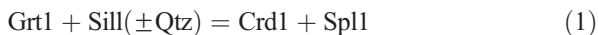
All samples are aluminous–magnesian granulites.

Table 7  
Representative chemical composition of plagioclase and mesoperthite

Samples	KJ-3	KJ-3	BJ-137-A	BJ-137-A	JB-05	X4-050-B	BJ-137-A	X4-050-B
Analysis no.	1	3	7	20	249	16	21	23
Remarks	Pl1(Core)	Pl1(Core)	Pl1(Core)	Pl1(Core)	Pl1(Core)	Pl1(Core)	Kfs/Opx2(Core)	Mp1(Core)
SiO <sub>2</sub>	59.44	59.38	58.67	58.92	55.95	59.05	63.61	63.92
TiO <sub>2</sub>	0.01	0.03	0.07	0.01	0.00	0.11	0.05	0.09
Al <sub>2</sub> O <sub>3</sub>	26.01	26.11	25.81	25.43	28.29	25.61	18.24	18.48
Cr <sub>2</sub> O <sub>3</sub>	0.00	0.00	0.12	0.06	0.02	0.10	0.00	0.08
FeO	0.02	0.07	0.02	0.12	0.14	0.04	0.01	0.06
MnO	0.00	0.00	0.00	0.00	0.00	0.12	0.00	0.00
MgO	0.00	0.00	0.00	0.00	0.00	0.00	0.03	0.01
CaO	6.75	7.53	7.73	7.30	10.38	7.43	0.12	0.05
Na <sub>2</sub> O	7.79	7.25	6.81	6.98	5.80	7.31	0.72	0.20
K <sub>2</sub> O	0.22	0.33	0.49	0.64	0.09	0.42	16.23	16.31
Sum	100.24	100.70	99.72	99.46	100.67	100.99	99.01	99.20
<i>Structural formulae based and 8 oxygens</i>								
Si	2.66	2.63	2.63	2.66	2.49	2.63	2.98	2.98
Al	1.32	1.36	1.38	1.35	1.48	1.34	1.01	1.02
Ti	0.00	0.00	0.01	0.00	0.00	0.01	0.00	0.00
Cr	0.00	0.00	0.01	0.00	0.01	0.01	0.00	0.00
Fe <sup>3+</sup>	0.00	0.01	0.00	0.00	0.01	0.01	0.00	0.00
Fe <sup>2+</sup>	0.00	0.00	0.01	0.01	0.00	0.00	0.01	0.00
Mn	0.00	0.00	0.00	0.00	0.00	0.01	0.00	0.00
Mg	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Ca	0.32	0.34	0.35	0.35	0.49	0.49	0.01	0.00
Na	0.67	0.63	0.60	0.61	0.50	0.50	0.07	0.02
K	0.01	0.01	0.03	0.01	0.01	0.01	0.97	0.97
Ab	66.80	62.30	61.51	62.47	50.03	62.53	6.65	1.87
An	32.00	35.80	35.60	36.09	49.46	35.09	0.63	0.25
Or	1.20	1.30	2.89	1.44	1.90	2.38	92.71	97.88

XH-050-B is garnet-bearing charnockitic granulite sample and the others are aluminous–magnesian granulites.

et al., 1990) and were formed by the following univariant reaction in the FMAS system (Fig. 3a, b).



We interpret the orthopyroxene and K–feldspar coronae around biotite (Bt1) (Fig. 3f) as the products of the multivariant biotite dehydration–melting reaction in the KFMASH system as suggested by Mohan and Windley (1993):



The reaction coronae were formed after ductile deformation, since they surround primary biotite (Bt1), which is oriented parallel to the rock foliation. This dehydration reaction leads to the presence of water during metamorphism, and can be used to estimate the value of  $a_{\text{H}_2\text{O}}$  based on the experimental studies of Hoffer and Grant (1980).

Cordierite porphyroblasts (Crd2) around spinel crystals were formed through the retrograde decompression reaction (Fig. 3c, g), which was studied by Mohan

and Windley (1993) and Prakash (1999) in the FMAS system:



Orthopyroxene (Opx3) plus cordierite (Crd3) symplectite around Grt1 was produced through decompression, and is inferred to record the FMAS divariant reaction 4 (Fig. 3d, e). Similar to symplectites found in the aluminous–granulites from the Central Highland Complex, Sri Lanka (Sajeev and Osanai, 2004), the symplectite orthopyroxene in Bahia granulites also shows an increase in Fe towards the rim (Table 3). The large volume change of this reaction makes it particularly appropriate for geobarometry (Harris and Holland, 1984a,b).



We suggest that Bt2 + Qtz symplectites (Fig. 3h) were formed by back-melting reactions during cooling, as proposed by Kriegsman (2001) in a similar case:

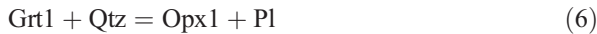




Leucosomes and even orthopyroxene-bearing S-type granites occur in the aluminous–magnesian granulites (Barbosa et al., 2004), showing that melt was formed probably from metapelites along the prograde path and its continued presence was responsible for the formation of biotite (Bt2) during cooling.

6.2. Garnet-bearing charnockitic granulite

The studied garnet-bearing charnockitic granulite from HG (Fig. 2) is the product of granulite facies metamorphism of a plutonic protholith. Equilibrium contacts are usually observed between the main mineral phases, although destabilisation of garnet is observed when it is in contact with quartz, which produces orthopyroxene plus plagioclase symplectites. Garnet megacrysts in contact with quartz are armoured by coronae of orthopyroxene (Opx1) and plagioclase (Pl1), which shows the breakdown of garnet following the silica-saturated FMAS reaction (Raith et al., 1997):



Fine-grained biotite+quartz symplectite was also formed substituting orthopyroxene. These microstructures can be interpreted as products of the multivariant KFMASH reaction:



Barbosa and Sabaté (2004) demonstrated that the rocks of the Jequié block, including HG, were originally in amphibolite facies, and were transformed

to granulites during the paleoproterozoic collision. Biotite and hornblende in these granulites are in disequilibrium with orthopyroxene (Barbosa, 1986), showing that melt must have been produced during prograde metamorphism, and may have persisted in small quantities in these rocks, thus being responsible for the formation of retrograde biotite Bt2 during cooling. On the other hand, retrograde biotite is also found around orthopyroxene and opaque minerals in shear zones formed during the uplift of the Jequié block, by influx of water. This water could also have invaded the garnet-bearing charnockitic granulite, thus forming Bt2 during cooling.

7. Thermobarometry

P–T conditions of the studied rocks were determined using GPT (Reche and Martinez, 1996) and THERMOCALC (Holland and Powell, 1998) softwares (Tables 8–10). The samples were those described in detail above: aluminous–magnesian granulite samples KJ-3, KJ-1, and BJ-137-A from close to Brejões Dome, and aluminous–magnesian granulite JB-05 and garnet-bearing charnockitic granulite XH-050-B away from the dome (Fig. 2).

They are rocks with preserved microstructures such as coronae and symplectites that record the decompression processes, yielding information about a number of stages along the retrograde P–T path. On the other hand, little evidence for the prograde P–T path remains. The regression method of Pattison and Bégin (1994) was

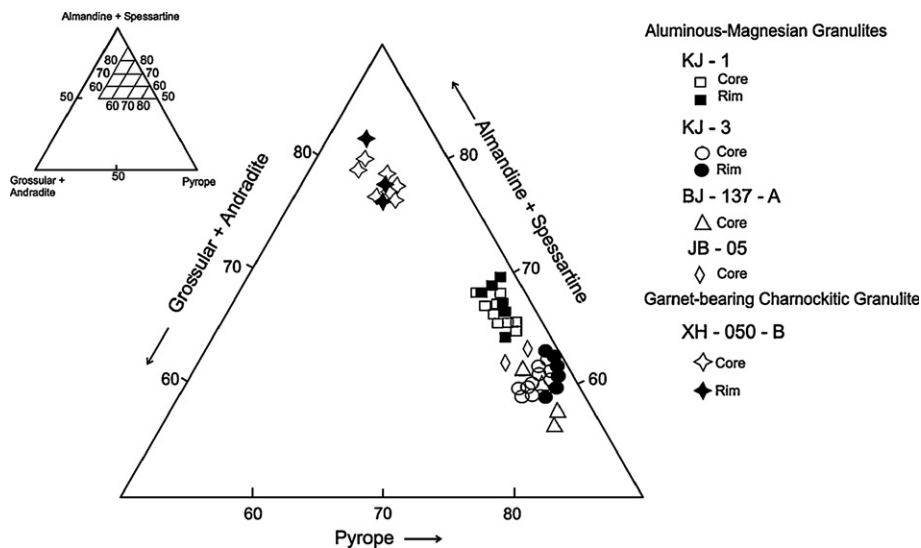


Fig. 4. Garnet compositions of garnet-bearing charnockitic granulite and Al–Mg granulite in the pyrope–(grossular + andradite)–(almandine + spessartite) diagram. All analysed garnet grains are Grt1.

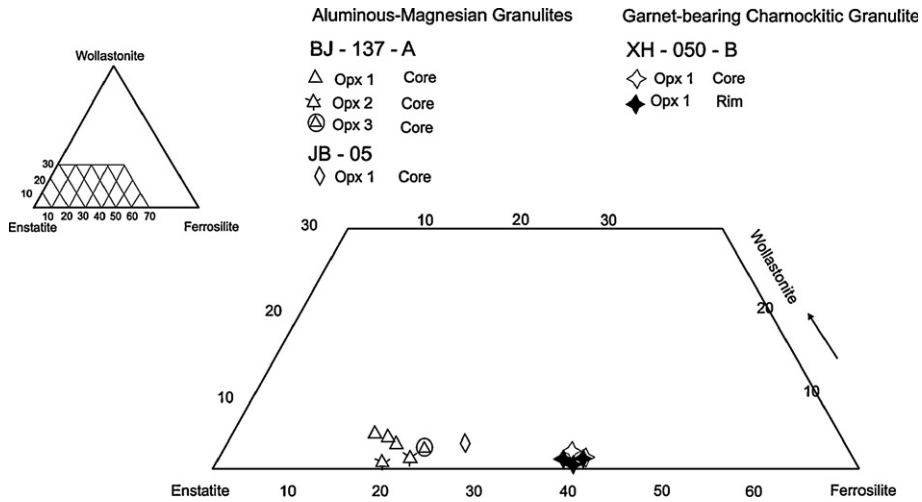


Fig. 5. Orthopyroxene compositions of garnet-bearing charnockitic granulite and aluminous–magnesian granulites in the wollastonite–ferrosilite–enstatite diagram.

applied to retrieve close-to-peak mineral compositions from which close-to-peak *P* and *T* estimates were obtained with the Al–orthopyroxene–garnet barometer of Harley and Green (1982), and the garnet–orthopyroxene geothermometer of Harley (1984), respectively, because they are based on the same set of experiments.

Temperatures of 900–1150 °C were determined using the Grt–Ilm (Pownceby et al., 1991) and Crd–Spl (Vielzeuf, 1983) geothermometers for the Al–Mg granulite samples containing spinel (KJ-3, BJ-137-A) close to the Brejões charnockite intrusion (Table 8). The calibration of Holland and Powell (1998) also gives high temperatures (900–1000 °C) for these samples, but indicates lower temperatures (850–870 °C) for the

garnet-bearing charnockitic granulite sample (XH-050-B) (Table 9) collected farther away from the border. The hercynite+quartz assemblage was formed at a temperature of at least 800 °C (Ellis et al., 1980; Tobi et al., 1985; Sandiford and Powell, 1986; Waters, 1991). The analysed spinel contains between 0.76 and 2.75 ZnO wt.%, in the same range as found in spinel from ultrahigh-temperature granulite belts (e.g. Dasgupta et al., 1995; Ouzegane and Boumaza, 1996), while the Al<sub>2</sub>O<sub>3</sub> content of 6.4 wt.% in orthopyroxene cores from the aluminous–magnesian granulite (Table 3) also falls within the range of Al<sub>2</sub>O<sub>3</sub> contents found in orthopyroxene cores from the typical ultrahigh-temperature granulite terrain in Enderby Land, Antarctica (Grew, 1982; Harley et al., 1990).

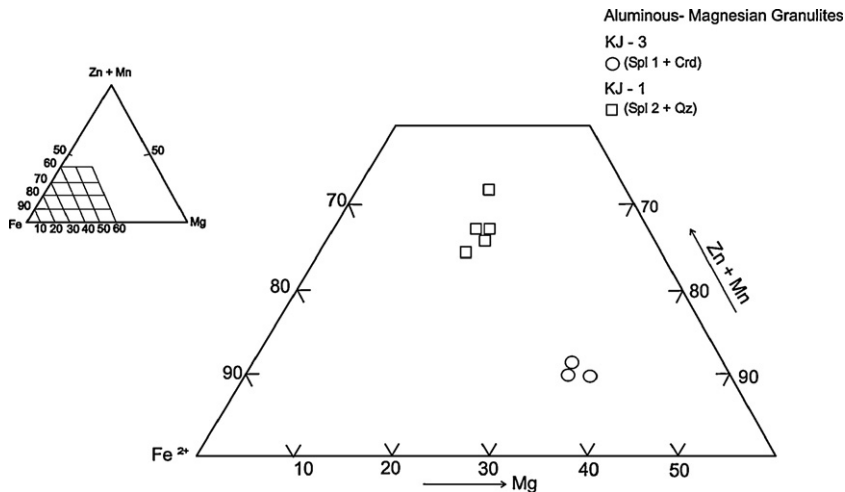


Fig. 6. Spinel composition of hercynite–quartz-bearing granulite in the Zn+Mn/Fe<sup>2+</sup>/Mg diagram.

Table 8

Geothermometry of the studied rocks, according to the method of Reche and Martinez (1996)

Geothermometry/samples	KJ-3	KJ-1	BJ-137 A	JB-05	XH-050-B
Brejões Dome	Samples near Dome			Samples far from Dome	
Reche and Martinez (1996)					
Gt–Crd (7–10 kbars)					
Thompson (1976)	777–836	734–769	721–737		
Holdaway and Lee (1977)	730–790	692–706	695–708		
Bhattacharya et al. (1988)	753–793	737–758	716–731		
Gt–Ilm (7–10 kbars)					
Pownceby et al. (1991)	919–1148	900–1100	991–1090		
Gt–Opx (7–10 kbars)					
Sen and Battacharya (1984)			786–809	790–810	657–692
Harley (1985)			724–741	710–780	617–649
Lee and Ganguly (1988)			800–819	800–820	697–729
Lal (1993)			674–696	700–710	690–719
Crd–Spl (7–10 kbars)					
Vielzeuf (1983)	818–896	970–902			

Most of calculated *P* values are between 5 and 8 kbar for all analysed samples, irrespective of the calibration used (Grt–Pl–Bt–Qtz; Grt–Pl–Sil–Qtz; Grt–Rt–Sil–Ilm–Pl–Qtz) (Table 10). Pressure estimates following Holland and Powell (1998) are closer to 8 kbar for both the aluminous–magnesian granulites (KJ-1), and the garnet-bearing charnockitic granulite (XH-050-B) (Table 9).

Rocks from the Brejões Dome area (re-)crystallised under low to intermediate pressure (5–8 kbar) granulite facies conditions. Temperatures determined in samples collected away from the dome are in the order of 850 °C, similar to that found elsewhere in the southern Bahia high-grade region (Barbosa, 1986, 1990). Samples from close to the dome, however, record higher temperatures of about 900–1000 °C which are typical of ultrahigh-temperature granulite conditions (Harley, 1998). This is confirmed by the *P–T* conditions calculated using the protocol of Pattison and Bégin (1994) for garnet–orthopyroxene pairs in aluminous–magnesian granulite (sample BJ-137-A). The calculated *P* is 12.0±2.0 kbar at *T* of 960±40–60 °C. The errors are after Harley (1984) who assumes ±0.1 as the maximum error in the calculated *K<sub>D</sub>*. The data suggest that the Brejões charnockite intrusion was an additional heat source which modified the regional granulite facies metamorphism.

### 8. Textural interpretation and *P–T* history

The high-temperature assemblages (Spl+Qtz) and (Opx1+Bt1+Grt1+Sill+Qtz) were formed, and the reaction Grt1+Sill (±Qtz)=Crd1+Spl1 occurred at about 7–8 kbar and 1100 °C and can be explained by using the FMAS system. The retrograde reactions Spl2+

Qtz=Crd2 and Grt1+Qtz=Opx3+Crd3, which mainly occurred during orogenic decompression and cooling, can also be explained by using the FMAS system according to Harris and Holland (1984a,b), Harley and Hensen (1990) and Waters (1991). For reaction Spl2+Qtz=Crd2 at high temperature we used *X<sub>Mg</sub>*=0.7. The partial pressure of oxygen (*fO<sub>2</sub>*) is a major factor controlling the stability of the assemblages and reactions (Hensen, 1986). The mineral reactions and the presence of ilmenite inclusions in garnet are consistent with *fO<sub>2</sub>* remaining low throughout their evolution.

On the other hand, reactions Bt1+Qtz=Opx2+Kfs+L and Opx2+Kfs+L=Bt1+Qtz can be explained by using the KFMASH system as suggested by Mohan and Windley (1993). The first multivariant dehydration

Table 9

Geothermobarometry of the studied rocks according to the method of Holland and Powell (1998)

Geobarometry/ samples	KJ- 3	KJ-1	BJ- 137-A	BJ- 137-A	JB- 05	XH- 050-B
Dome de Brejoes	Samples near Dome				Samples far from Dome	
Holland and Powell (1998)						
(Pl–Opx–Gt– Bt–Ilm–Fk)						
			1051 °C;			887 °C;
			7.5 Kb			7.1 Kb
			984 °C;			889 °C;
			8.2 Kb			7.3 Kb
			985 °C;			850 °C;
			8.2 Kb			7.3 Kb
			900 °C;			
			7.3 Kb			

Table 10

Geobarometry of the studied rocks, according to the method of Reche and Martinez (1996)

Geobarometry/samples	KJ-3	KJ-1	BJ-137-A	BJ-137-A	JB-05	XH-050-B
Brejões Dome	Samples near Dome				Samples far from Dome	
Reche and Martinez (1996)				(700–850 °C)		
Gt–Pl–Bt–Qtz (850–1100 °C)						
Hoisch (1990) (Mg)			7.6–9.0	5.4–6.1	4.6–5.9	6.7–8.2
Hoisch (1990) (Fe)			7.4–9.5	6.1–7.0	3.9–5.0	9.1–11.6
Gt–Pl–Sill–Qtz (850–1100 °C)				(700–850 °C)		
Newton and Haselton (1981)	7.6–9.9	6.8–9.1	7.1–9.3	6.0–6.8	5.4–6.9	
Hodges and Spear (1982)	5.7–8.1	5.1–7.1	5.3–7.5	4.4–5.0		
Ganguly and Saxena (1984)	7.4–8.9	6.7–8.9	7.2–9.6	5.4–6.6	5.0–6.5	
Hodges and Crowley (1985)	7.4–9.3	6.7–8.6	7.0–8.8	5.5–6.7	5.0–6.4	
Koziol (1989)	9.2–11.7	8.5–10.9	8.6–11.1		7.0–9.0	
Koziol and Newton (1989)	8.3–10.6	7.6–9.7	7.9–10.0	6.2–7.6	6.0–7.7	
Gt–RT–Sill–Ilm–Pl–Qtz				(700–850 °C)		
Bohlen et al. (1983)	6.8–7.6	6.8–7.6	8.7–10.0	4.7–6.8		
Bohlen et al. (1983)	4.8–6.1	5.1–4.8	6.7–7.9	6.2–7.0		
Bohlen and Liotta (1986)	7.3–9.8	6.9–8.3	6.8–8.2	4.1–5.2		

reaction, valid for  $P_{\text{H}_2\text{O}}=P_{\text{Total}}$ , moves to a higher temperature, probably due to a reduced water activity that could be caused by an influx of  $\text{CO}_2$  during granulitization (Newton, 1992). Corroded blebs of biotite within orthopyroxene suggest that this reaction did occur. Textural evidence for the second hydration in the form of biotite plus quartz symplectites around orthopyroxene shows that the first reaction was reversed to produce the second reaction.

## 9. Tectonic setting and conclusions

Low to intermediate pressure and high temperature conditions usually prevailed in Archean/Paleoproterozoic high-grade metamorphic terrains. Many granulite terrains, especially those with hercynite + quartz-bearing granulite, cooled from very hot transient conditions. The most generally applicable models involve a combination of magmatic advection and lithospheric thinning (Waters,

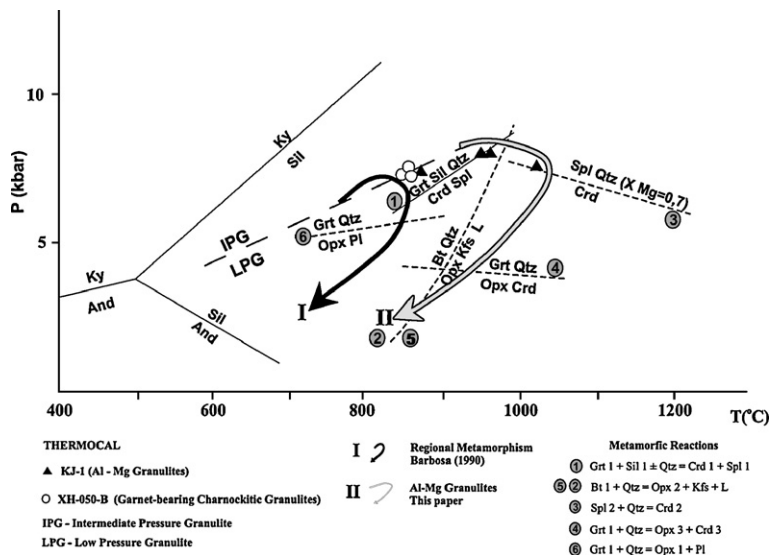


Fig. 7.  $P$ - $T$  path of the Brejões Dome area. The higher  $T$  path was obtained with the samples near the Brejões Dome and the lower  $T$  path (Barbosa, 1990) is from the rocks away from the dome. Reactions Grt1 + Sill ± Qtz = Crd1 + Spl1, Spl2 + Qtz = Crd2 and Grt1 + Qtz = Opx3 + Crd3, observed in rocks of this study, are from Harris and Holland (1984a,b), Harley and Hensen (1990), Waters (1991) and Mohan and Windley (1993). The reactions are calculated using the thermodynamic data base of Holland and Powell (1998). IPG-intermediate pressure granulite and LPG-low pressure granulite (Green and Ringwood, 1972). Aluminium silicates triple point is from Holdaway (1971).



1991). Thermal aureoles around magmatic intrusions contain high-grade granulite with hercynite+quartz paragenesis (Berg, 1977). The hercynite+quartz association, found in some granulite xenoliths in basaltic rocks, also indicates metamorphic temperatures in excess of 900 °C (Vielzeuf, 1983).

Clockwise *P–T* paths are characteristic of geotectonic regions dominated by collision/obduction. Thickening of the continental crust is followed by rapid uplift

and erosion, in general at plate margins (England and Richardson, 1977).

The metamorphic path of the granulite region of southern Bahia (Fig. 2) is clockwise, with low/intermediate pressures (5–8 Kb) at high temperature (850–870 °C) (Barbosa, 1990) (Fig. 7). It is admitted that this metamorphism was the consequence of crustal thickening, due to the collision with Archean segments (Fig. 8a, b, c), followed by rapid uplift and erosion

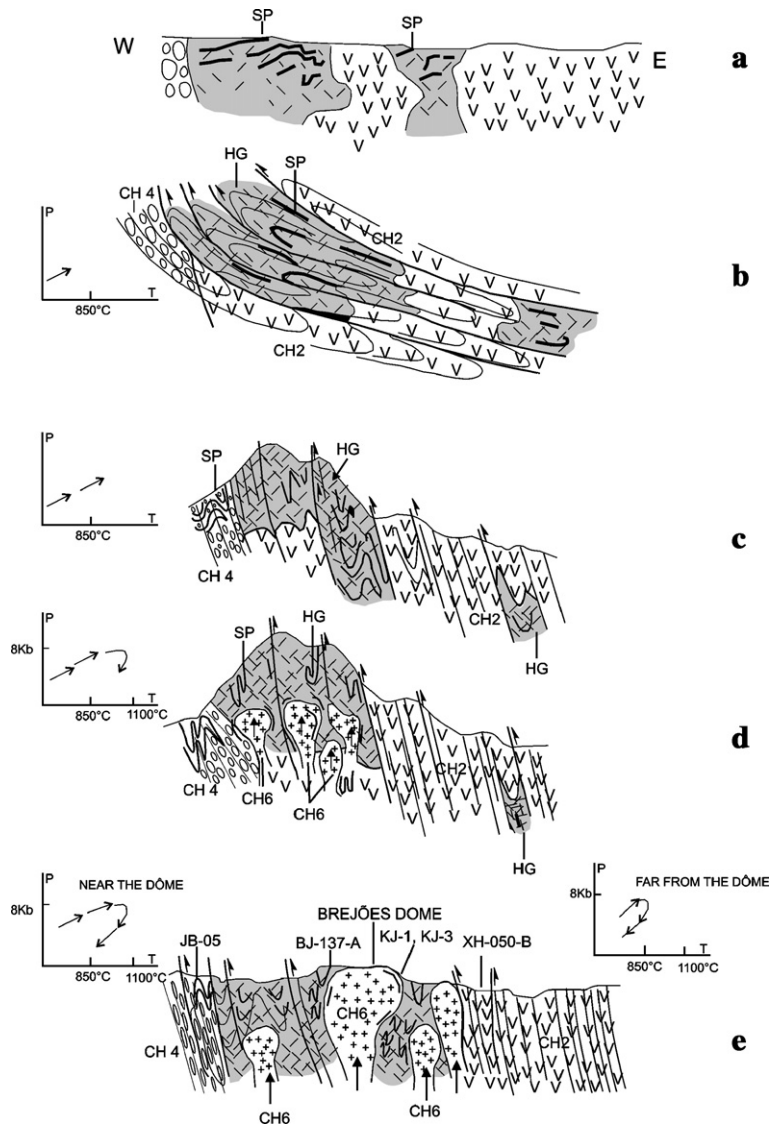


Fig. 8. Geotectonic evolution and *P–T* paths of the Brejões Dome area. a—Intrusion of enderbite–charnockitic granulites protoliths, dated at 2.7–2.8 Ga, within the Archean crust with supracrustal rocks (SP). b and c — Paleoproterozoic crustal thickening resulting in granulite facies metamorphism ( $\approx 2.04$  Ga) and deformation: first phase of sub-horizontal tectonics (F1), resulting in large west vergent recumbent folds (e.g. Irajuba nappe, in CH4, Fig. 3); second coaxial phase (F2); third phase (F3), represented by vertical shear zones, sub-parallel to (F2) axial planes. d — Late charnockite intrusions (CH6), such as the Brejões and Santa Inês domes, leading to an increase of thermal gradient around 2.04 Ga. e — Present situation of the Brejões Dome area and sampling localities close to the Brejões Dome (low/intermediate pressure and ultrahigh-temperature metamorphism) and away from the dome (low/intermediate pressure and high temperature).

(Barbosa, 1990; Figueiredo and Barbosa, 1993). The present study corroborates the clockwise  $P$ – $T$  path already recorded for the rocks of the high-grade region of southern Bahia. Textural relationships record reactions such as  $\text{Grt1} + \text{Sil} \pm \text{Qtz} \Rightarrow \text{Crd1} + \text{Spl1}$  (Reaction 1), and  $\text{Bt1} + \text{Qtz} = \text{Opx2} + \text{Kfs} + \text{L}$  (Reaction 2), which indicate temperature increase (and/or low  $a_{\text{H}_2\text{O}}$ ). These reactions were followed by reactions  $\text{Spl1} + \text{Qtz} = \text{Crd2}$  (Reaction 3),  $\text{Grt1} + \text{Qtz} = \text{Opx3} + \text{Crd3}$  (Reaction 4),  $\text{Opx1} + \text{Kfs} + \text{L} = \text{Bt2} + \text{Qtz}$  (Reaction 5) and  $\text{Grt1} + \text{Kfs} + \text{L} = \text{Bt2} + \text{Qtz}$ , which record reduction of  $P$  and  $T$  (Fig. 7). On the other hand, samples from close to the contact of the Brejões Dome also define a clockwise path at a higher temperature than that recorded in samples away from the dome. It is suggested that the intrusion of the CH6 Brejões charnockite diapir, generated by underplating of the lower crust by magmas which provoked its partial melting, was responsible for the local temperature rise above the peak regional metamorphic temperature (Fig. 8d, e). The hercynite–quartz assemblage in supracrustal rocks surrounding the Brejões dome suggests a very high temperature ( $>1000$  °C) for the CH6 magmas which would have induced partial melting in adjacent supracrustal rocks, forming garnet–cordierite-bearing granitic melts. Finally, it is uncommon to find evidence for the reaction  $\text{Bt1} + \text{Qtz} = \text{Opx2} + \text{Kfs} + \text{L}$  occurring in both directions, and we believe that this was possible here because the intrusion of the CH6 Brejões charnockite diapir raised the temperature in the area.

## Acknowledgements

We acknowledge a fellowship from Project 381/02-CAPES-COFECUB, the financial support from CBPM—Companhia Baiana de Pesquisa Mineral and CNPq—Conselho Nacional de Desenvolvimento Científico e Tecnológico, Brazil.

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